

A STUDY OF PHASE STABILITY IN INVAR Fe-Ni ALLOYS  
OBTAINED BY NON-CONVENTIONAL METHODS

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It is known that thermodynamic equilibrium in Fe-Ni alloys, in composition at temperatures below 450°C is difficult to achieve because of the diffusion rate at low temperatures. One of the ways in which we can transformation which may be responsible for invar behavior is to i) materials of similar composition obtained by non-conventional methods allow the enhancement of diffusion at temperatures where atomic mobility is the laboratory time scale; ii) materials which have been treated for long periods of time (geological time scale) in the same temperature range, such as meteorites. In this context we have studied the phase stability of mechanically alloyed powders, beam mixed multilayers and meteorites.

## I - INTRODUCTION

The invar alloys, which are based on the composition Fe-36% Ni, have a zero thermal expansion coefficient over a substantial temperature range. High temperature properties and parameters (lattice parameter, electrical resistivity, magnetization and elastic moduli) show anomalies which have been observed experimentally. These observations indicate that there is a phase transition in the Fe-Ni phase system at the invar composition. Several authors suggested a low temperature miscibility gap to explain the anomalies and have proposed a variety of phase diagrams for the Fe-Ni system.

The Fe-Ni phase diagram proposed by Romig and Goldstein in 1968 [1] showed two major phases,  $\alpha$  and  $\beta$ , and a two phase region. Using this phase diagram, for a typical invar alloy, heat treatment at 800°C. The alloy will enter the two phase region during the cooling process.

The low temperature portion of the Fe-Ni phase diagram was recently proposed [2, 3] based on meteoritical evidence and electron irradiation of Fe-Ni alloys. The authors concentrated on the composition range from 0 to 50% Ni on temperatures below 450°C (fig. 1). Both the stable and metastable phase boundaries are defined. They observed an asymmetrical miscibility gap in the metastable region below 390°C and is caused by the presence of a tricritical point. One can apply this phase diagram to Fe-Ni alloys of 36 wt% Ni. This phase diagram is not simple and the low temperature phase transformations are not still well understood.

The major experimental difficulty in studying these alloys is the slow rate of the Fe-Ni system at these low temperatures. As cooling occurs

coefficient of Ni decreases, for example from  $6.05 \times 10^{-16} \text{ cm}^2 \text{ s}^{-1}$  at  $500^\circ \text{C}$  to  $1 \times 10^{-21} \text{ cm}^2 \text{ s}^{-1}$  at  $300^\circ \text{C}$ . At  $300^\circ \text{C}$  it takes more than 10<sup>4</sup> years for one atomic jump to occur.

One of the ways in which we can study phase transformations which are responsible for invar behavior is to investigate materials of similar composition prepared by non-conventional methods which are known to enhance diffusion, allowing equilibrium to be reached in short times. So, to accelerate the attainment of an equilibrium-like state we studied samples prepared by non-conventional methods as: i) Fe-Ni alloys in the form of fine particles prepared by mechanical alloying; ii) Fe-Ni multilayers in a very thin modulation ion bombarded with noble gases; iii) Meteorites which contain Fe-Ni alloys slowly cooled after solidification; iv) Fe-Ni alloys in natural bodies for millions of years.

## II - MECHANICAL ALLOYED FE-NI ALLOYS

Mechanical alloying (MA) is a new technique of combining metals by the high energy ball milling of elemental powders in a high energy ball mill under an inert atmosphere. It circumvents many of the limitations of conventional alloying and creates a wide variety of metals or metal-non metal composites that are difficult or impossible to produce by other means.

We used MA as an alternative method to prepare Fe-Ni alloys, since it produces grain refinement and produces great amount of lattice defects. Therefore we expect that MA can greatly enhance the diffusivity of the powders. The details are described in ref. [4] and in this Proceedings. The longest annealing time was 90h and that powder was subsequently annealed at  $350^\circ \text{C}$  for 350t times. The

composition of the powder after milling was checked by EDS, indicating no detectable difference from the starting powder.

The phase separation process during the subsequent heat treatment by X-ray diffraction (XRD), Mössbauer spectroscopy (MS) and magnetic measurements.

The XRD patterns of the starting and milled (10 to 90h) powders show that after 10h milling a Ni<sub>40</sub>Fe<sub>60</sub> alloy with fcc structure is formed as indicated by the disappearance of the diffraction peaks from the pure metals. The diffraction pattern of the powder milled 90h does not show significant difference from that of the powder milled 10h in both, the structure and the linewidth. The effective crystallite size is similar for the two milled powders.

The MS of the starting powder and the ones milled for different times show an alloying process in the initial stage of the milling showing several peaks indicating a mixture of phases with different compositions. This is due to interdiffusion, i.e., Fe diffuses in Ni matrix and Ni diffuses in Fe matrix, the coexistence of Fe-rich and Ni-rich alloys. This behavior is characteristic of a mechanically alloyed FeCr system. After 10h milling, an homogeneous alloy is formed and further milling does not change the alloy structure. This is confirmed by the spectra of the powders after 10h milling. All of them have the same hyperfine field (31T) and similar hyperfine field distribution, typical of a Ni<sub>40</sub>Fe<sub>60</sub> alloy (40%).

The powder milled for 90h was submitted to 350°C annealing at different times (fig. 2). After 20h annealing, segregation starts to occur with the appearance of a non-magnetic phase, a singlet (IS=-0.15 mm/s) that corresponds to the fcc  $\gamma$ -phase with 130%. After 100h annealing, segregation becomes more pronounced.

with formation of a magnetic component ( $H=20T$ ) superimposed to the non-magnetic phase. This is an intermediate stage in which Fe concentration is low non-magnetic phase but is higher than that in the initial alloy. After the non-magnetic phase increases and the magnetic spectrum shows a complex structure. The spectrum can be fitted adding a component with  $H=29T$  at  $100\text{ mm/s}$ . These parameters are typical of the Fe-Ni 50-50 ordered phase  $L1_0$ . This seems to be a sign of formation of the ordered phase in the annealing process.

Using MA as an alternative method we obtained a defective nano-structured Fe-Ni disordered alloy which submitted to long time annealing at  $350^\circ\text{C}$  showed phase segregation with formation of phases with different Ni composition [4].

### III - ION - BEAM MIXED Fe-Ni MULTILAYERS

It is known that irradiation with energetic particles is an effective way of enhancing atomic diffusion in metals, thereby reducing the time required to reach phase equilibrium. Extensive studies have shown that Fe-Ni invar alloys undergo phase segregation after enough irradiation (neutrons or electrons) to enhance atomic diffusion. It has also been found that the alloys with composition in the range of the invar anomalies occurs are those with the greatest response to irradiation.

One of the alternative ways of achieving a state closer to thermodynamical equilibrium (which for this system means atomic ordering and phase segregation), is to use ion bombardment in Fe-Ni multilayers.

We investigated the effect of noble gas (He, Ne, Ar and Xe) ion bombardment on Fe-Ni multilayers with nominal composition  $Fe_{50}Ni_{50}$  through CEMS.

The Fe-Ni multilayers were prepared using e-gun source in a vacuum system at the Institute for Chemical Research, Kyoto University during the deposition was better than in these conditions it was produced multilayers with total thickness of 1020 Å and a very thin modulation (0.18 Å Ni) or a nominal composition. The ion irradiations were done at the HVEE 400-kV ion implantor of the Institute of Physics, Universidade Federal do Sul [6].

Typical CEMS spectra of films irradiated using Ne ions (70 keV) are shown in fig. 3. The as-deposited sample as well as the irradiated sample, display only the typical sextet of bcc alpha-Fe. However, it is clearly seen the formation of Fe-Ni phases with different Ni composition one corresponding to a magnetic phase atomically ordered - Fe<sub>50</sub>Ni<sub>50</sub> - (QS=0.15 - 0.20 mm/s) and another one corresponding to a non-magnetic phase at.%. With increasing doses the ordered phase increases up to 18% while the magnetic component presents a remarkable enhancement up to 10% accounting for 40% of the spectrum at this dose. An increase of dose did not lead to any alteration of the spectra, suggesting that there is

The same effect but less effective was observed by irradiation with similar doses. The two phase region (ordered + non-magnetic), obtained with He irradiated samples, is the same already observed in particle irradiated samples [7] and meteorites [8,9], in which it has been considered as the equilibrium

The non-magnetic phase formed by Ne irradiation shows an interesting behavior, vanishing completely when further irradiated with Xe. If we change the order of irradiation, first Xe and then Ne, phase segregation does not occur, the magnetic phase is not obtained and the CEMS spectra displays the same

of hyperfine fields produced by Xe irradiation, showing that the phase Xe predominates over the others.

Irradiation of Fe-Ni multilayers, in the invar region, with a series of ions (He, Ne, Ar and Xe) allowed us to evaluate the formation/stability of phases formed by ion irradiation [6]. Our results can be interpreted as evidence that for lighter ions (He, Ne) phase separation is obtained and equilibrium liquid system is achieved, whereas for heavier ions (Ar, Xe and Kr) the mixed phase is predominant.

#### IV - METEORITES

Meteoritic metal contains a unique complicated microstructure. It has been observed that in slowly cooled meteorites their metallic microstructure is formed by a series of complex phase transformations occurring below 400°C. These microstructures can be observed in the 3 main groups of meteorites: a) in chondritic meteorites = chondrites (from the mantle of parent bodies); b) in intermetallic meteorites (from mantle/core interface or from collision mixing); and c) in Fe-rich meteorites (from Fe-rich Fe-Ni cores of parent bodies).

The study of meteoritic metal is an attempt to determine the phase diagram experimentally. Meteoritic metal is basically an Fe-Ni alloy with 5 to 60 at% Ni with small amounts (< 1 wt%) of Co, P, S, and C. Because meteorites have cooled slowly over millions of years (1 to 1.000 million years) in parent bodies, meteoritic metal contains a characteristic structure which can be completely duplicated in the laboratory due to the slow diffusion at low temperatures. Therefore, meteorites are useful as indicators of the phase transformation which occur in Fe-Ni alloys. The microstructure

temperature phase transformation products in meteoritic metal are stony-iron and iron meteorites. Differences in the microstructure are a function of cooling history at low temperature.

It should be noted that the metallic phases of meteorites, produced by low temperature phase transformations, are sub-micron in size due to the low diffusivity of the system. Therefore for this particular phase equilibrium the X-ray diffraction technique is of little use due to the crystallographic complexity of the resultant phases. Mössbauer spectroscopy, on the other hand has played a major role in the meteorite work. In particular Mössbauer measurements gave the first evidence for the existence of the superposition of a ferromagnetic phase with a  $\text{Fe}_{50}\text{Ni}_{50}$  phase (tetraenite,  $H = 29\text{T}$ ;  $QS = 0.20\text{ mm/s}$ ) and a non-magnetic phase with  $\approx 10\%$  (fig. 4).

The non-magnetic phase usually referred to as "paramagnetic phase" was recently reported by Rancourt and Scorzelli [10] as a possible equilibrium phase in the Fe-Ni system. We propose that this phase, with estimated composition  $\text{Fe}_{50}\text{Ni}_{50}$  (a low spin Fe-Ni phase), which in synthetic irradiated alloys and meteorites always occurs in a fine epitaxial intergrowth with ordered FeNi. Since it is seen in coexistence with FeNi having different degrees of atomic order (depending on the sample) it has been proposed that this phase occurs in close microstructural association with FeNi.

This phase is only observed by Mössbauer spectroscopy because it has the same lattice parameters as FeNi (ordered FeNi) and is therefore indistinguishable from FeNi. These phases are therefore readily observable as a distinct phase by TEM or X-ray diffraction.



So, the proposed tetraenite intergrowth that is a common state in slow cooled iron meteorites, is present in metal particles of chondrites which has also been observed in synthetic irradiated alloys, mechanically alloyed and ion irradiated thin films, can be considered as indicative of the equilibrium state of Fe-Ni at the invar composition.

## FIGURE CAPTIONS

Fig. 1 - FeNi phase diagram proposed by Reuter et al [2] based on the iron meteorite structure and electron irradiated alloys.

Fig. 2 - Mössbauer spectra at room temperature of a 90% Ni powder Fe submitted to annealing for the indicated times.

Fig. 3 - CEM spectra of Fe-Ni multilayers: a) as deposited; and Ne i: doses b)  $5 \times 10^{15} \text{ Ne/cm}^2$  c)  $10^{16} \text{ Ne/cm}^2$  d)  $5 \times 10^{16} \text{ Ne/cm}^2$  e)  $10^{17} \text{ Ne/cm}^2$

Fig. 4 - Mössbauer spectrum at room temperature of the Santa Catharina

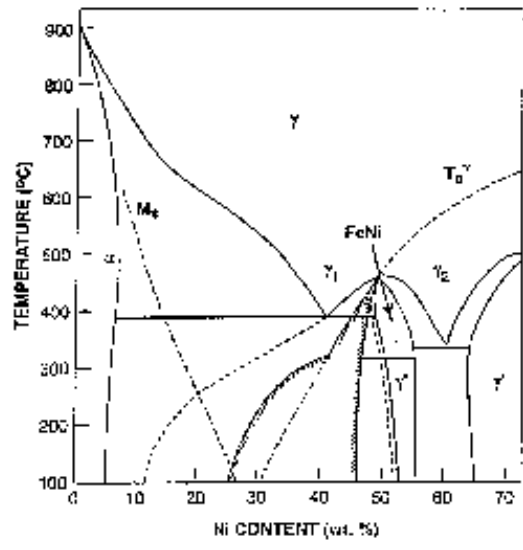


Figure 1

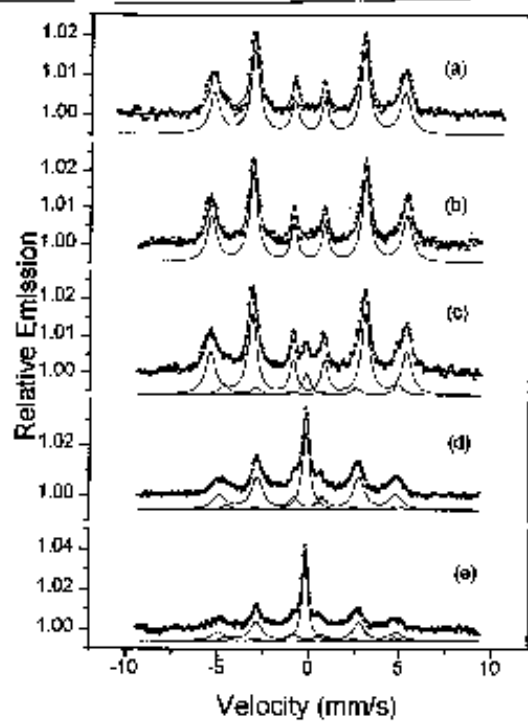


Figure 2

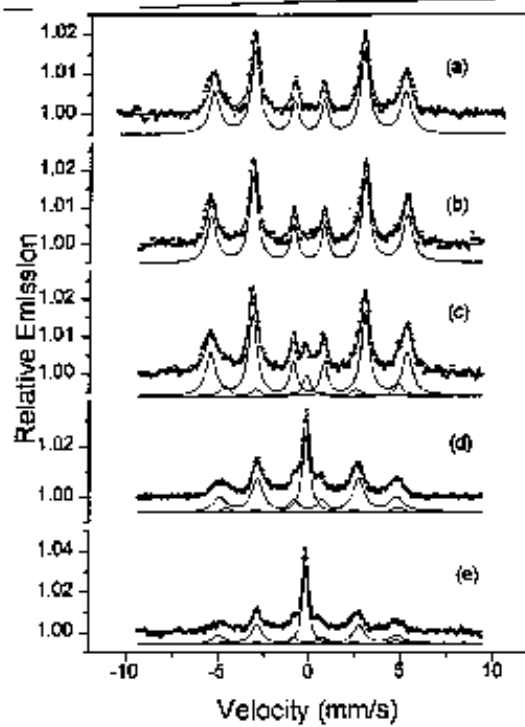


Figure 3

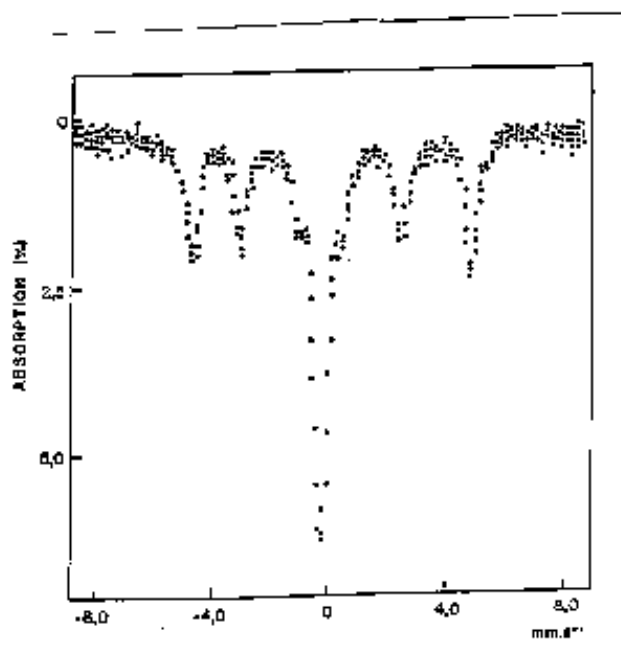


Figure 4

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